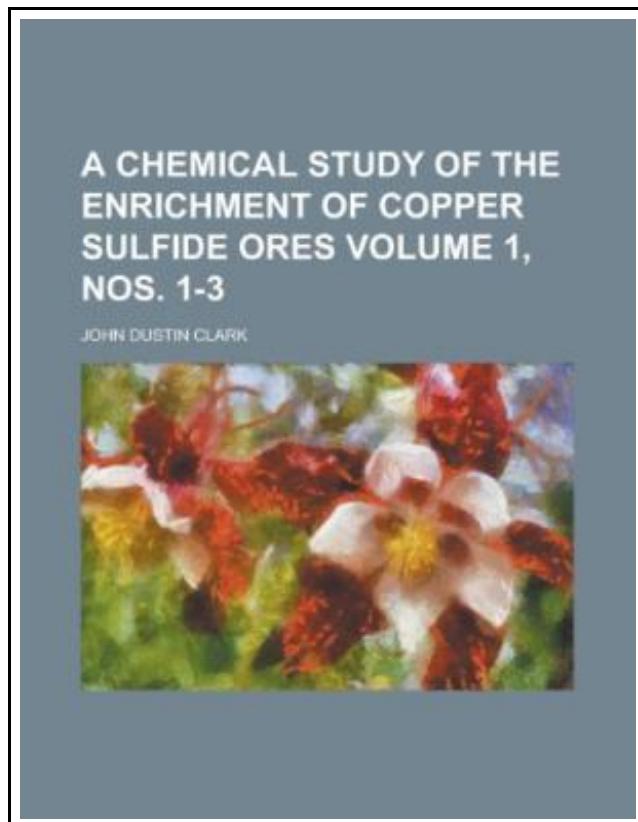


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Reviews

Merely no terms to spell out. We have read through and i also am confident that i will gonna read yet again again in the future. You will not sense monotony at anytime of your own time (that's what catalogs are for about should you question me).

(Pasquale Larkin I)

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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1913 edition. Excerpt: .mine waters. Lindgren1v in summarizing Emmon s work mentions the possibility of solution as chlorides where nascent chlorine has played an important role, this chlorine having been produced in accordance with the question: $MnO_2 + 4 HC_1 = MnCl_4 + H_2O$ He gives instances of increased solution of gold when manganese dioxide has been added to a solution containing very dilute hydrochloric acid. The above five methods of oxidation seem to cover the principles involved. One will find, however, mention of many possibilities of oxidation and solution which combine two or more of the above methods. Considerable work has been done in attempting to discover the order in which different minerals oxidize. No two investigators have succeeded in getting the same results1. In all the investigations made the investigators have failed to consider conditions under which the oxidations took place and their results put the chemist in mind of the old Geoffrey and Bergmann affinity tables. The essential factors entering into the problem of oxidation are; ore, metal, country rock, fissuring, porosity, climate, water level, rainfall, topography, geological age and history of the deposit. These have to be considered in the examination of any particular mine. PRECIPITATION OF THE METALLIC SULFIDES. Various methods have been suggested by which the metals in descending solutions of metallic salts can be precipitated as sulfides. The following tried and untried means seem to cover the possibilities of such precipitations: (1) Substitution because of solution tension of some mineral...



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